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Production, characterization and *In vitro* antioxidant activities of exopolysaccharide from *Weissella cibaria* GA44

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1. Introduction

Many LAB are able to produce EPS which are known to improve the physical properties of fermented food products. The increase in demand for the production of polysaccharides by microbes for use in food and pharmaceutical industries has been enormously researched on (Imran et al., 2016; Kajala et al., 2015; Sajna et al., 2013). Exopolysaccharides are economically significant because they can impart functional effects to food and also confer beneficial health effects on human when consumed (Welman & Maddox, 2003). Polysaccharides with beneficial health effects are widely used in the food industry as gelling, thickening, stabilizing, or emulsifying agents. They play a major role in the rheology and texture of the fermented food products (Dilna, Surya, Aswathy, Varsha, & Sakthikumar, 2015).

Several functional EPS which are produced from LAB species isolated from various fermented sources have displayed both antioxidant and probiotic properties (Pan & Mei, 2010; Zhang et al., 2013; Abdhul et al., 2014). Antioxidants have received enormous attention by researchers/scientists due to the increased number of diseases caused by the formation of free radical and/or reactive

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ABSTRACT

Exopolysaccharide (EPS) producing lactic acid bacteria (LAB) strain GA44 isolated from *gari* (a Nigerian traditional fermented cassava mash) was identified with 16S rRNA gene sequencing as *Weissella cibaria*. The EPS produced mainly composed of glucose and rhamnose sugar units and its molecular mass was about 2.8 \times 10⁵ Da. The fourier-transform infrared (FT-IR) spectrum analysis showed characteristics absorption peaks indicating the presence of -OH, C=O and C-H groups. Central composite design (CCD) and response surface methodology (RSM) were used to study the interactive effect of the significant factors on EPS yield. Under optimum conditions, the predicted maximum EPS produced 4.88 g/L while the actual experimental value was 4.80 g/L. The *in-vitro* antioxidant activities of the EPS showed good scavenging effects on superoxide anion radical and hydroxyl radical. The characteristics of this EPS isolated from food source will be useful in food industries and could also serve as a potential antioxidant. © 2017 Elsevier Ltd. All rights reserved.

oxygen species (Emerit, Edeas, & Bricaire, 2004; Kris-Etherton et al., 2002). Reactive oxygen species (ROS) are chemically reactive species containing oxygen, which easily react with other molecules in a cell. They are formed as a natural by-product of the normal metabolism of oxygen and play important roles in cell signaling and homeostasis. Accumulation of these ROS in cells, or their increase as a result of environmental stress may cause damage to DNA, RNA, and proteins, and may even leads to cell death (Melov et al., 2000). Exopolysaccharides from LAB are referred to as natural antioxidants and have been reported safe as they played roles in the maintenance of human health, prevention and treatment of diseases, they may also serve as possible substitutes for the synthetic antioxidants because of the toxicity effect the latter displays (Li et al., 2013; Liu, Ooi, & Chang, 1997).

Researchers have reported that *Weissella* species can produce significant amount of EPS (Badel, Bernardi, & Michaud, 2011; Kajala et al., 2015). *W. cibaria* have been isolated from plants and vege-tables, and also from different indigenous fermented foods of Nigeria such as gari, fufu, ogi (Adesulu-Dahunsi, Sanni, & Jeyaram, 2017). *Weissella* sp. have the ability to produce significant amounts of non-digestible oligosaccharides and extracellular polysaccharides which can be used as prebiotics or for other applications in food industries (Baruah, Maina, Katina, Juvonen, & Goyal, 2017; Malang, Maina, Schwab, Tenkanen, & Lacroix, 2015). Some LAB strains utilizes the fermentable sugars for the







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biosynthesis of EPS. Sugars such as glucose, lactose, sucrose and mannose serve as carbon sources (Patel, Majumder, & Goyal, 2012; Xu, Ma, Wang, Liu, & Li, 2010).

The LAB producing strain, growth medium and culture conditions may influence the EPS production and their monomer compositions (De Vuyst & Degeest, 1999; Degeest, Vaningelgem, & De Vuyst, 2001). Similarly, the quantity of EPS produced by different strains of LAB varies considerably (Badel et al., 2011: Laws & Marshall, 2001). Response surface methodology has recently been used for modeling and optimization of processes such as fermentation media, process and cultivation conditions for EPS production (Imran et al., 2016; Raza, Makeen, Wang, Xu, & Qirong, 2011). In this paper, W. cibaria GA44 isolated from gari was screened and identified to produce appreciable amounts of exopolysaccharide in modified MRS-sucrose media. The EPS was purified, the chemical properties determined and its functional groups were detected by infrared spectrophotometry. Response surface methodology was used to determine the optimal conditions for the EPS yield. The antioxidant activities were also investigated using various in vitro assays such as 1, 1-diphenyl-2-picryl-hydrazyl (DPPH) free radical scavenging activity, reducing power assay, superoxide anion and hydroxyl radical scavenging activity in order to understand the potential usefulness of the EPS from the indigenous fermented food as a natural antioxidant.

2. Materials and methods

2.1. Microorganism and reagents

Exopolysaccharide producing LAB strain *W. cibaria* GA44, isolated from *gari* was identified according to its morphological characteristics and 16S rRNA sequence analysis. The nucleotide sequence was submitted to the GenBank under the nucleotide accession number; KU892397. The culture was grown in Man Rogosa and Sharpe (MRS) medium at 30 °C for 48 h and maintained on the agar at 4 °C. T-series dextrans were purchased from Pharmacia Co. ltd. (Uppsala, Sweden), DPPH, trichloroacetic acid (TCA), trifluoroacetic acid (TFA), bovine serum albumin (BSA), phenol, concentrated sulfuric acid, pyrogallol, Folin-Ciocalteuvd reagent, methanol, ferric chloride, potassium ferricyanide, glucose, galactose, fructose, mannose, p-glucuronic acid, rhamnose, xylose and standard antioxidant compound ascorbic acid were all purchased from Sigma chemical co. Itd. (St. Louis, USA). All other reagents used were of analytical grade.

2.2. Partial purification of the EPS

The EPS was extracted, isolated and purified following the method described by Savadogo et al. (2004), with slight modifications. *Weissella cibaria* GA44 was incubated at 30 °C for 24 h in MRS broth, 100 mL of the culture was inoculated into 900 mL modified MRS (*m*-MRS) broth (glucose was replaced with sucrose as carbon source at a concentration of 20 g/L) and was incubated in an incubator shaker. After 24 h, the cultures were heated at 100 °C for 10 min to inactivate enzymes potentially capable of polymer degradation and the cells were removed by centrifugation at 12,000×g for 15 min at 4 °C. The supernatant was precipitated with double volume chilled ethanol, shaken vigorously and centrifuged at 5000×g for 30 min at 4 °C. The precipitate was dried at 50 °C.

Finally, the precipitate was dissolved in ultrapure water, and was again precipitated with double volume of chilled ethanol, this step was repeated three times and dialyzed against distilled water for 2 day at 4 °C using 10 kDa dialysis membrane with the changing of the water twice daily and then lyophilized. Freeze-dried EPS was used for further characterization.

2.3. Estimation of total sugar and protein content

The total sugar content of the EPS was determined by phenolsulfuric acid method (Dubois, Gilles, Hamilton, Rebers, & Smith, 1956). The total protein content was determined by Folin's Lowry method (Lowry, Rosebrough, Farr, & Randall, 1951).

2.4. Molecular mass determination

The molecular mass of the EPS was determined on high performance liquid chromatography (HPLC) system (Agilent 1100) equipped with a TSK-GEL G3000SWxl column (7.5×300 mm, Tosoh Corp. Tokyo, Japan) and a refractive index detector (RID). The column was eluted with 0.1 M Na₂SO₄ solution at a flow rate of 0.8 mL min⁻1. Molecular mass was estimated from the standard graph which was plotted using standard dextrans (Sigma–Aldrich, USA).

2.5. Fourier transform infrared (FT-IR) spectroscopy

The functional groups of the purified EPS were detected by FT-IR spectroscopy. The sample pellet was prepared by grinding a mixture of the EPS (1 mg) with 100 mg of dry KBr powder, followed by pressing the mixture into the mold. FTIR spectra was recorded on a Thermo Nicolet 6700 instrument in the ranges of $400-4000 \text{ cm}^{-1}$.

2.6. Monosaccharide composition analysis of EPS

The monosaccharide composition of the EPS was analyzed. The EPS was hydrolysed by adding 5 mg of the EPS sample to 0.5 mL of 6 N trifluoroacetic acid (TFA) at 100 °C for 2 h, the TFA was then removed by evaporation under N₂ gas. After hydrolysis, 2 μ l of the hydrolyzate EPS sample and seven standard sugars (glucose, galactose, fructose, mannose, D-glucuronic acid, rhamnose and xylose) were spotted each onto a silica gel coated aluminum TLC plates. The mixed solvent system n- butanol, ethanol and water (50:30:20 (v/v/v)) were used for separation of carbohydrates and standards and were developed in the TLC chamber, the TLC plate was dried and developed by dipping it into anisaldehyde-sulfuric acid reagent and heating in an oven at 110 °C for 30 min (Seo et al., 2014).

2.7. Effect of culture conditions and the media compositions

Initial screening for maximum EPS production was carried out by detecting the influence of different carbon and nitrogen sources. Twenty-four (24) hours old culture was inoculated into the media components at different pH, cultivation time and temperatures. The EPS yield and cell growths were evaluated.

2.8. RSM optimization for enhanced EPS production

Based on the results obtained from the screening of the culture parameters and media compositions, the CCD was used to find the optimal values for these significant variables, cultivation time, sucrose concentration, yeast extract, pH and to understand the relationship between these factors and EPS production. A multiple regression analysis of the data was carried out with the statistical Design-Expert[®], version 8.0.7.1 package (Stat Ease Inc. Minneapolis, USA).

2.9. Assay of antioxidant activities

2.9.1. DPPH radical scavenging assay

The DPPH radical scavenging activity was measured by using the method described by Rai et al. (2011). Briefly, 2.0 mL DPPH radical solution (0.16 mM) was added to 1.0 mL sample solution (0.5–4 mg/mL), and then 2.0 mL of deionized water was added. The mixture was mixed vigorously and incubated at 37 °C in the dark for 30 min. The absorbance was measured at 517 nm against the blank. Methanol with DPPH served as a positive control. The sample blank was prepared by replacing DPPH with methanol.

The DPPH radical (%) scavenging activity was estimated by the following equation:

$$1 - \frac{[Asample - Ablank]}{Acontrol} \times 100$$

2.9.2. Reducing power potential

The reducing power potential of the EPS sample at different concentration levels (0.5–4 mg/mL) was measured using the method described by Balakrishnan et al. (2011). To 100 μ L of the EPS sample, 900 μ L of phosphate buffer (0.2 M, pH 6.6) and 900 μ L of 1% potassium ferric cyanide are added, and was incubated at 50 °C for 20 min. 900 μ L of 10Q% TCA was added, mixed and centrifuged at 5000 \times g for 15 min and 900 μ L of the supernatant solution was mixed with 900 μ L of distilled water and 900 μ L of 0.1% ferric chloride. The solution was mixed and the absorbance was taken at 700 nm.

2.9.3. Superoxide radical scavenging activity

Super oxide scavenging property of the EPS sample was determined according to the method described by Balakrishnan et al. (2011). To 0.3 mL of the EPS sample at different concentration levels (0.5–4 mg/mL), 2.6 mL of 50 mM phosphate buffer (pH 8.2) was added and to this 90 μ L of freshly prepared 3 mM pyrogallol dissolved in 10 mM HCl was added. Then the absorbance was measured from 0 min to 10th min at 325 nm. To 0.3 mL of distilled water, 2.6 mL phosphate buffer (50 mM (pH 8.2)) was added and this served as blank.

The superoxide scavenging activity (%) was estimated by the following equation:

$$1 - \frac{|A_{10} - A_0|}{C_{10} - C_0} \times 100$$

 A_0 and A_{10} are the absorbance of the EPS sample at 0 and 10 min C_0 and C_{10} are the absorbance of the control at 0 and 10 min.

2.9.4. Assay of hydroxyl radical scavenging activity

The hydroxyl radical scavenging effect of the EPS sample was measured with the Fenton reaction. Briefly, the reaction mixture contained 1.0 mL of brilliant green (0.435 mM), 2.0 mL of FeSO₄ (0.5 mM) and 1.5 mL of H_2O_2 (3.0%, w/v). One (1) mL of the sample at various concentrations (0.5–4 mg/mL) were added and was incubated at room temperature for 20 min, and the absorbance was measured at 624 nm (Balakrishnan et al., 2011).

The hydroxyl radical scavenging activity (%) was estimated by the following equation:

$$\frac{[A_{\rm S}-A_0]}{A-A_0}\times 100$$

 $A_S = Absorbance \ of \ the \ sample$

 $A_0 = Absorbance of the control$

 $\mathbf{A}=\mathbf{A}\mathbf{b}\mathbf{s}\mathbf{o}\mathbf{r}\mathbf{b}\mathbf{a}\mathbf{n}\mathbf{c}$ and Fenton reaction

2.10. Statistical analysis

All experiments were performed in triplicates and the results represented by their mean \pm SD (standard deviation). Tests of significant differences were determined by Duncan's Multiple Range Test at (P < 0.05).

3. Results and discussion

3.1. Estimation of total sugar and protein content

The total sugar content in EPS was estimated to be 84.60%. Wang, Zhao, Tian, Yang, and Yang (2015) and Imran et al. (2016) also



Wavenumber (cm⁻¹)

Fig. 1. FTIR spectrum of EPS produced by Weissella cibaria GA44.

Table 1

Effects of carbon source, nitrogen sources and yeast extract on cell growth and EPS production by Weissella cibaria GA44.

Culture conditions	Growth (OD 600 nm)	EPS (g/L)					
Carbon sources (20 g/L)							
Glucose	1.462 ± 0.08	2.87 ± 0.02					
Sucrose	1.565 ± 0.06	3.82 ± 0.03					
Lactose	1.432 ± 0.04	2.21 ± 0.14					
Galactose	1.501 ± 0.01	3.01 ± 0.17					
Sucrose conc. (g/L)							
10	1.435 ± 0.02	3.00 ± 0.12					
20	1.668 ± 0.04	3.80 ± 0.08					
30	1.559 ± 0.07	2.90 ± 0.06					
40	1.413 ± 0.09	1.10 ± 0.04					
50	1.332 ± 0.01	0.87 ± 0.07					
Nitrogen sources (25 g/L)							
Yeast extract	1.765 ± 0.08	3.64 ± 0.02					
Beef extract	0.897 ± 0.15	2.51 ± 0.04					
Tryptone	1.332 ± 0.06	3.01 ± 0.06					
Peptone	1.257 ± 0.01	3.22 ± 0.10					
Yeast extract conc. (%, w/v)							
1.0	1.340 ± 0.02	2.18 ± 0.11					
1.5	1.323 ± 0.07	2.32 ± 0.06					
2.0	1.408 ± 0.04	3.07 ± 0.01					
2.5	1.442 ± 0.03	3.60 ± 0.11					
3.0	1.556 ± 0.02	3.53 ± 0.11					

OD600

EPS

4

3

2

1

0

8

Data are means \pm SD of triplicates.

2

1.8

1.6

1.4

1.2

1

0.8

0.6

0.4

0.2

0

6

6.5

7

OD 600nm

reported high sugar content greater than 90.00% from EPS produced from LAB. The EPS exhibited a low total protein content of approximately 1.12%. Similarly, other researchers have reported low protein content from purified EPS samples. Liu, Wang, Lin, & Lin, 2002 reported 2.30% protein from EPS obtained from the fermentation of kefir grains in soya milk.

3.2. Molecular mass of EPS

The molecular mass of EPS was determined by HPLC with size exclusion column. Based on the calibration curve of the elution time of various standard dextrans (Fig. S1 of Supplementary material), the molecular mass of EPS was estimated to be 2.8×10^5 Da.

3.3. Fourier transform infrared (FT-IR) spectroscopy

FT-IR is an effective tool for detecting the structural and functional groups in EPS (Wang, Zhang, Zhang, & Li, 2008). Fig. 1 showed the absorption bands of EPS-GA44 as determined by FTIR spectrum. The broad stretching peak of 32287 cm⁻¹ observed describe the characteristics of a carbohydrate ring and the absorption of this region revealed that EPS contains rounded trait typical of hydroxyl groups which shows that the EPS sample is a polysaccharide. The bands in the 2901 cm⁻¹ region are due to C–H-





Fig. 2. Effects of (A) initial pH (B) temperature and (C) incubation time on cell growth and EPS production by Weissella cibaria GA44. Data are means \pm SD of triplicates.

Table 2a

Central Composite Design matrix of four variables showing the predicted and actual values of production of exopolysaccharides from Weissella cibaria GA44.

Run	X ₁	X ₂	X ₃	X4	EPS Yield (g/L)	
	Cultivation time (h)	Sucrose concentration (g)	Yeast extract (%)	pH	Actual	Predicted
1	48.50	20.00	2.00	8.00	4.41	4.44
2	49.00	20.00	2.50	7.50	4.80	4.88
3	47.50	18.00	2.60	7.00	4.80	4.78
4	46.01	16.00	3.00	6.00	4.31	4.29
5	46.08	24.00	1.70	6.00	4.35	4.37
6	46.00	24.00	2.40	8.00	4.81	4.79
7	49.00	24.00	2.60	6.00	4.22	4.30
8	46.00	17.00	1.93	5.00	4.01	4.00
9	48.70	24.00	2.05	6.00	4.35	4.37
10	48.84	19.96	2.52	5.60	4.80	4.82
11	48.36	22.93	2.48	7.55	4.83	4.85
12	48.22	21.91	2.38	6.00	4.60	4.62
13	48.78	20.02	2.70	6.05	4.43	4.50
14	46.32	20.11	2.55	7.51	4.86	4.87
15	46.08	18.35	1.86	6.07	3.98	4.00
16	45.69	16.39	2.47	7.51	4.55	4.46
17	47.01	19.05	2.58	6.41	3.99	4.02
18	46.59	18.77	1.78	6.45	4.05	4.00
19	46.08	22.35	2.38	7.02	4.32	4.35
20	47.69	22.81	3.00	6.30	4.07	4.10
21	47.01	23.31	2.80	7.85	4.11	4.10
22	48.59	18.38	1.00	6.55	3.87	3.90
23	46.72	16.81	3.00	7.14	4.05	4.10
24	47.17	23.41	2.65	6.57	4.12	4.10
25	48.16	21.37	2.86	6.63	4.09	4.05
26	48.32	16.21	1.92	6.80	4.11	4.12
27	45.00	20.04	2.51	7.32	4.67	4.70
28	46.00	20.42	1.26	5.67	3.22	3.40
29	46.91	21.62	2.77	6.04	3.75	3.80
30	47.66	22.40	3.00	7.11	4.19	4.20

stretching vibration, and the bands around 1650 cm⁻¹ correspond to C=O and carboxyl group. The absorption band at 1344 cm⁻¹ represents the stretching vibration of the esters and ether groups. The strong absorption at 1154 cm⁻¹ was dominated by glycosidic linkage (C–O–C)-stretching vibration. The presence of different functional groups revealed a typical heteropolymeric nature of the EPS sample. The configuration of polysaccharides fraction determined using FT-IR spectroscopy have been reported by other researchers (Liu et al., 2010; Imran et al., 2016).

3.4. Monosaccharide composition analysis of EPS

The EPS sample showed two distinguishable sugar spots on the TLC plate. Based on their retention force (*Rf*) values when compared with the standard sugars, they were identified as glucose and rhamnose. The presence of different sugar moieties suggests that the EPS is an heteropolysaccharide. Baruah et al. (2017) reported the EPS produced by *Weissella cibaria* RBA12 was composed of only glucose unit. Fukuda et al. (2009) also reported that EPS from

Table 2b

Analysis of Variance (ANOVA) of quadratic model for production of exopolysaccharide in Weissella cibaria GA44.

Source	Sum of Squares	df	Mean Square	F Value	p-value Prob > F	
Model	1.35	14	0.74	7.07	0.0003	Significant
A-Cultivation time	0.023	1	0.023	1.73	0.2088	
B-Sucrose conc.	3.375	1	3.375	0.025	0.8769	
C-Yeast extract	0.085	1	0.085	6.27	0.0243	
D- Ph	1.504	1	1.504	0.11	0.7439	
AB	0.013	1	0.013	0.93	0.3497	
AC	0.083	1	0.083	6.08	0.0262	
AD	0.019	1	0.019	1.39	0.2565	
BC	7.656	1	7.656	0.56	0.4644	
BD	0.039	1	0.039	2.87	0.1108	
CD	1.806	1	1.806	0.13	0.7205	
A ²	0.030	1	0.030	2.20	0.1586	
B ²	0.48	1	0.48	35.06	< 0.0001	
C^2	0.61	1	0.61	44.99	< 0.0001	
D ²	0.27	1	0.27	19.90	0.0005	
Residual	0.20	15	0.20			
Lack of Fit	0.12	10	0.12	0.67	0.7246	not significant
Pure Error	8.087	5	5.13			Ū.
Cor Total	1.55	29				

R-Squared = 0.8602; Adequate Precision = 8.691.

L. fermentum TDS030603 contained glucose and galactose using TLC technique.

3.5. Effect of culture conditions and media composition on EPS production

The effect of different carbon and nitrogen sources on EPS yield and cell growth was investigated in MRS broth (Table 1). Sucrose was found to be the best carbon source for EPS production $(3.82 \pm 0.03 \text{ g/L})$. Examining the different concentration levels of sucrose (10-50 g/L), the optimum EPS yield and cell growth was $3.80 \pm 0.08 \text{ g/L}$ (OD600 1.668 ± 0.04) at 20 g/L. Moreover, decrease in cell growth and EPS yield were observed at sucrose concentration more than 20 g/L. Similarly, Kim, Robyt, Lee, Lee, and Kim (2003) and Sarwat, Aman, and Ahmed (2008) reported that increase in sucrose concentration results in high EPS yield. The effect of different nitrogen sources on EPS production by *W. cibaria* GA44 showed yeast extract as the best nitrogen source with the optimum EPS yield and highest cell growth of 3.64 ± 0.02 , OD600 1.765 ± 0.08 . This may be due to the presence of larger quantities of free amino acids and short peptides in the yeast extract. At various concentration of yeast extract, maximum EPS was produced at 2.5% ($3.60 \pm 0.11 \text{ g/L}$). EPS production by LAB is greatly influenced by nitrogen sources (Ismail & Nampoothiri, 2010; Liu et al., 2009; Wang et al., 2010).

The effect of initial pH, different temperature and cultivation time for EPS production and cell growth by *W. cibaria* GA44 for pH (6.0–8.0), temperature (20–40 °C) and cultivation time (12–96 h) were analyzed in MRS broth (Fig. 2). The optimal pH, temperature and cultivation time for cell growth and EPS production were 7.0, 30 °C and 48 h, respectively with the corresponding cell growth (OD600 1.71 \pm 0.02, 1.61 \pm 0.05 and 1.62 \pm 0.02) and EPS (g/L) production (3.63 \pm 0.10, 3.39 \pm 0.12 and 3.58 \pm 0.34). The quantity of the EPS produced is dependent on the microorganisms and their



Fig. 3. Response surface 3D plots and resulting contour plots of the effects of four significant variables on EPS yield for *Weissella cibaria* GA44 (A) Sucrose concentration and cultivation time (B) Yeast extract and cultivation time (C) Cultivation time and pH (D) Sucrose concentration and yeast extract (E) Sucrose concentration and pH (F) Yeast extract and pH.

culture conditions such as temperature and pH (Wang et al., 2010). The optimum temperature ranges for EPS yield from LAB have been reported to be 25–35 °C. Ismail and Nampoothiri (2010) and Kanmani et al. (2011) have reported the maximum EPS production by *L. plantarum* MTCC 9510 and *S. phocae* P180 at 35 °C. Optimum pH values ranging between 6.5 and 7.5 have been reported (Kanmani et al., 2011; Raza et al., 2011). Researchers have also reported that during prolonged cultivation time, the yield of EPS decreased gradually (Hallemeersch, De Baet, & Vandamme, 2002, pp. 407–429; Sarwat et al., 2008).

3.6. RSM for EPS production

To determine the optimum combination of the significant factors (sucrose, yeast extract and incubation time) in the medium, the CCD of RSM was employed. There was a total of thirty

experimental runs based on the model generated by the software, each in triplicates (Table 2a). The experimental data were statistically analyzed using the Fischer's statistical test for analysis of variance (ANOVA) and the 3D graphs were designed. Table 2b showed that the ANOVA for the model, the model p-value of 0.0003 was significant. The model was further confirmed significant with a model F-value of 7.07. The adequate precision measures the signal to noise ratio and the ratio of 8.691 obtained indicates an adequate signal that the model can be used to navigate the design space since a ratio greater than 4 is desirable for the precision of a model. The lack-of-fit term was greater than 0.05 which is non-significant relative to the pure error, in this case nonsignificant lack of fit is good. Hence, X_3 , X_1X_3 , X_2^2 , X_3^2 , X_4^2 are the significant model terms. The polynomial equation derived on the basis of the experimental factors, and the interactions among the factors is shown below:



Fig. 3. (continued)

$$\begin{split} Y &= +4.79 + 0.031 X_1 + 3.750 X_2 + 0.060 X_3 - 7.917 X_4 \\ &\quad + 0.028 X_1 X_2 + 0.072 X_1 X_3 - 0.034 X_1 X_4 + 0.022 X_2 X_3 \\ &\quad - 0.049 X_2 X_4 - 0.011 X_3 X_4 - 0.033 X_1{}^2 - 0.13 X_2{}^2 \\ &\quad - 0.15 X_3{}^2 - 0.099 X_4{}^2 \end{split}$$

where Y = is the EPS yield (g/L).

 X_1 , X_2 , X_3 and X_4 are the coded values of the test variables: cultivation time (h), sucrose concentration (g/L), yeast extract (%) and pH. The response of the RSM was shown as 3D response surface graphs and the combinations of the two factors selected keeping the other constant (Fig. 3). The curvatures' nature of 3D surfaces gave good interaction between sucrose concentration and cultivation time, yeast extract and cultivation time, cultivation time and pH, sucrose concentration and yeast extract, sucrose concentration and pH, yeast extract and pH. The optimal values of the tested variables was cultivation time of 49.00 h; sucrose concentration of 20.00 g/L; yeast extract of 2.50% and pH of 7.50. Under these conditions, the maximum predicted yield of EPS was 4.88 g/L and its experimental (actual) yield was 4.80 g/L. Several researchers have studied the production of EPS in varying media and production conditions (Macedo, Lacroix, Gardner, & Champagne, 2002; Shene & Bravo, 2007). The results obtained in this present study showed



Fig. 3. (continued).



Fig. 4. In vitro antioxidant activities of EPS of Weissella cibaria GA44 (A) Scavenging effect of EPS on DPPH (B) Reducing power of EPS (C) Scavenging effect of EPS on superoxide anion radical and (D) Scavenging effect of EPS on hydroxyl radical. The results are represented as mean ± SD of triplicates.

that the mathematical model was suitable for the simulation of EPS production.

3.7. Assay of antioxidant activities

We assayed the antioxidant potential of the partially purified EPS-GA44. The EPS demonstrated significantly moderate antioxidant capacities in a dose-dependent manner in various *in vitro* models such as DPPH radical scavenging activity, reducing power assay, superoxide anion and hydroxyl radical scavenging activity and was compared with the control ascorbic acid.

3.7.1. DPPH radical scavenging activity

Fig. 4A showed the DPPH radical scavenging capacity of partially purified EPS from *W. cibaria* GA44 and the standard compound ascorbic acid. The scavenging activity of the EPS and ascorbic acid increases as their concentration increases, i.e. at low and high concentrations; 0.5 mg/mL (19.8% and 38.7%) and 4 mg/mL (48.9% and 82.1%) were obtained for the EPS and ascorbic acid respectively. Li et al. (2014) reported dose-dependent DPPH free radical scavenging activity of crude EPS from *L. helveticus* MB2-1. Similarly,

other researchers have reported strong and concentrationdependent antioxidant activities in EPS from LAB (Xu et al., 2010; Zhang et al., 2013).

3.7.2. Reducing power assay

During the assay of reducing power, the capability of antioxidant compounds to reduce Fe³⁺/ferricyanide complex to its ferric form was monitored by absorbance of Perl's Prussian blue formation at 700 nm (Chung, Chang, Chao, Lin, & Chou, 2002). The reducing power of EPS-GA44 increases at higher EPS concentration (Fig. 4B). The EPS revealed its potential antioxidant activity, but relatively lower than ascorbic acid which is the standard antioxidant used.

3.7.3. Superoxide anion scavenging activity

Superoxide anions are active free radical precursors that can react with biological macromolecules and have the potential to cause tissue damage (Pan & Mei, 2010). The scavenging activity of the EPS and ascorbic acid on superoxide radicals are shown in Fig. 4C. At the different concentrations tested (0.5–4 mg/mL), the scavenging activity of the EPS and ascorbic acid increased with increase in concentration. At 4 mg/mL concentration, the

scavenging activity of the EPS and the ascorbic acid was 77.1% and 83.1% respectively.

3.7.4. Hydroxyl radical scavenging activity

Hydroxyl radicals are responsible for the oxidative injury of biomolecules (Gülçin, 2006). The scavenging activity on hydroxyl radical of the EPS and ascorbic acid are shown in Fig. 4D. The scavenging effects of EPS and ascorbic acid increases with increase in their concentrations. At the concentration of 4 mg/mL, the scavenging capability of the ascorbic acid was 83%; however, the activity of the EPS reached up to 88%. This result indicates that the EPS has stronger scavenging activity on hydroxyl radical compared to that of the standard antioxidant used.

4. Conclusion

Exopolysaccharides from LAB have potential for development as functional food ingredients or food additives having both economic and health benefits. In the present work the optimum factors and culture conditions for production of EPS by lactic acid bacterium *Weissella cibaria* GA44 were identified. *In-vitro* antioxidant assay showed strong antioxidant properties, especially scavenging of superoxide anions and hydroxyl radicals. These results suggest that EPS from *Weissella cibaria* GA44 may be a better alternative natural source with potential antioxidant activity, and as well as healthy food grade adjunct culture.

Conflict of interest

Authors declare that there are no conflicts of interest whatsoever.

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Appendix A. Supplementary data

Supplementary data related to this article can be found at http://dx.doi.org/10.1016/j.lwt.2017.09.013.

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