## International Journal of Chemical and Process Engineering Research

2015 Vol. 2, No. 3, pp. 30-43 ISSN(e): 2313-0776 ISSN(p): 2313-2558 DOI: 10.18488/journal.65/2015.2.3/65.3.30.43 © 2015 Conscientia Beam. All Rights Reserved.



# MODELING AND OPTIMIZATION OF TRANSESTERIFICATION OF BENISEED OIL TO BENISEED METHYLESTER: A CASE OF ARTIFICIAL NEURAL NETWORK VERSUS RESPONSE SURFACE METHODOLOGY

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# ABSTRACT

In this research work, statistical approach (ANN and RSM) were used to optimize the transesterification of beniseed oil to beniseed methyl ester (BME). Analyses of an heterogeneous catalyst (Mangifera indica powdered) obtained from unripe Mangifera indica peels showed that the powder consist macro elements such as K (59.85%), Si (30.53%), Cl (4.58%), Al (3.05%) and Ca (1.05%) and micro elements such as P (0.196%), S (0.593%), Mn (0.043%), Fe (0.037%), Zn (0.008%), Rb (0.042%) and Sr (0.032%). ANN predicted optimal condition for Beniseed methyl ester produced was  $X_i = 60.0$  min,  $X_2 = 1.0$  wt.%,  $X_3 = 57$ °C and  $X_i = 6.0$ . The predicted BME (94.40% (w/w)) under this condition was validated to be of 93.80 % (w/w). Meanwhile, RSM predicted 94.20% (w/w) at the following condition  $X_i = 62.0$  min,  $X_2 = 0.9$  wt. %,  $X_3 = 60$  °C and  $X_i = 6.5$  was validated as 92.80 % (w/w). The results obtained showed the superiority of ANN over RSM owing to its higher values of predicted value, RMSE, AAD, R<sup>2</sup> and R<sup>2</sup><sub>A44</sub>. The fatty acid profile and the physicochemical properties of the BME indicated that, BME can serve as alternative fuel for conventional diesel.

**Keywords:** Optimization, Response surface methodology, Artificial neural network, Fatty acid profile, Physicochemical properties, Beniseed methyl ester.

# **Contribution**/ **Originality**

This study contributes in the existing literature to the process conversion of oil to diesel. This study uses new estimation methodology in the use of heterogeneous catalyst obtained from agricultural waste. This study originates new way of using statistical software (ANN and RSM) to improve the yield of biodiesel. This study is one of very few studies which have investigated

the use of heterogeneous catalyst to produced biodiesel. The paper contributes the first use of *Mangifera indica* (Mango) peels as heterogeneous base catalyst. The paper's primary contribution is finding that biodiesel can be obtained by the use of biomass waste. This study documents economic impact of using *Mangifera indica* (Mango) peels as heterogeneous base catalyst to produce biodiesel.

## 1. INTRODUCTION

As per the literature survey, the use of statistical software to optimize process conditions is receiving more attention day by day. Classical modeling techniques, such as response surface methodology (RSM) is an important tool which has been used extensively in laboratory research as well as industrial research, it major advantage is the capability to reduce the number of standard experimental runs, required to provide classically acceptable results [1]. In view of this, Betiku and Adepoju [2] worked on methanolysis optimization of sesame oil to biodiesel and fuel quality characterization using RSM. Jeong, et al. [3] carried out research on optimization of transesterification of animal fat ester using RSM. In the same vein, [4] worked on the biodiesel production from crude cottonseed oil using RSM. Statistical approach to the optimization of citric acid production using filamentous fungus Aspergillus niger grown on sweet potato starch hydrolyzate using RSM was carried out by Betiku and Adesina [1]. Tiwari, et al. [5] applied the same tool to optimized process of biodiesel production from *jatropha* oil. Statistical approach to alcoholysis optimization of sorrel (hibiscus sabdariffa) seed oil to biodiesel and emission assessment of its blends was carried out by Betiku, et al. [6]. Meanwhile, Adepoju, et al. [7] carried out worked on an innovative approach to biotransformation of benzaldehyde to L-PAC via free cells of Saccharomyces cerevisae in the presence of  $\beta$ -cyclodextrin using the same software. In another work, Adepoju, et al. [8] worked on optimization of benzyl alcohol production via biotransformation of benzaldehyde using free cell of saccharomyces cerevisiae in presence the  $\beta$ cyclodextrin using RSM. Guan and Yao [9] also make used of RSM in optimization of viscozyme L-assisted extraction of oat bran protein.

ANN is a learning system based on a computational technique that can simulate the neurological processing ability of the human brain and can be applied to quantify a non-linear relationship between connecting factors and actual responses by means of iterative training of data obtained from a designed experiment [10]. ANNs show superiority as a modeling technique for data sets showing non-linear relationships, and thus for both data fitting and prediction abilities [10, 11]. It has been use to solve myriads of problems in the field of medicine, metrology, neurology, biology, phycology, science, mathematics and engineering [12]. Ghaffari, et al. [13] worked on performance comparism of neural network training algorithms in modeling of biomodal drug delivery, Gueguim Kana, et al. [14] optimized biogas production from sawdust using ANN whereas Rajendra, et al. [15] applied the same tool to optimized the pretreatment

process parameters for biodiesel production, Ebrahimpour, et al. [16] also apply ANN in optimization studies of lipase production while Adepoju and Olawale [17] optimize transesterification of chrysophyllum albidium seed oil to chrysophyllum albidium oil biodiesel using artificial neural network. However, Betiku and Ajala [18] compared the performance of ANN and RSM in modeling and optimization of *thevetia peruviana* (yellow oleander) oil biodiesel synthesis via musa paradisiacal (plantain) peels, it was reported that ANN offers a promising outlook in the estimation of the optimum variables for biodiesel production, Adepoju, et al. [19] also compared the performance of ANN and RSM for achieving desire benzene alcohol in the biotransformation of benzaldehyde using free cells of *Saccharomyces cerevisae* and the effect of  $\beta$ -cyclodextrin and reported that ANN methodology presents a better alternative than the RSM model meanwhile, Dasai, et al. [20] compared the tools to work on fermentative production of *scleroglucan*, and established that using RSM and ANN along together showed a clear superiority over using each of them alone, Bosco Kana, et al. [21] compared ANN and RSM in modeling and optimization of citric acid production by AspergillusnigerMCBN297 and reported that RSM provided good insight on parameters interaction.In this study, an effort was made to optimize the reaction variables (reaction time, reaction temperature, heterogeneous catalyst amount and methanol / oil molar ratio) for transesterification of Sesamum indicum oil biodiesel synthesis via Mangifera indica (Mango) peels as heterogeneous base catalyst while comparing the performance of ANN and RSM.

## 2. MATERIALS AND METHODS

## 2.1. Materials

The seeds of *Sesamum indicum* were collected from Kabba, Nigeria. The dirty seeds were washed to remove the adherent dirty, sun dried for six days and then winnowed to remove the foreign matter (i.e. chaffs and grit). The cleaned seeds were then grinded into powdery using grinding machine. The *M. indica* (unripe) used in this study was obtained from Landmark University farm, Omu Aran. All chemical and reagents used were of analytical grades made by GFS Chemicals, Inc., 867 McKinley Ave., Columbus OH 43223 (99.7-100%) and BDH Analar Ltd., Poole England (99%) and supplied by FINLAB Nig. Ltd.

## 2.2. Methods

## 2.2.1. Beniseed (Sesamum Indicum) Oil Extraction

A 250 ml Soxhlet extractor apparatus with n-hexane as a solvent was used for oil extraction. At first, the thimble of apparatus was charged with a known weight of the powder seed placed in a muslin cloth. The solvent containing part of the apparatus was filled with a known volume of nhexane, fixed to the end of the thimble, and a condenser was tightly fixed at the bottom end of the extractor. The whole set up was placed in a temperature controlled heating mantle and heatedup at temperature of 68 - 70 °C. Excess solvent in the extracted oil was recycled and recovered by distillation. The oil obtained was stored in a five (5) litre keg for biodiesel production (methyl ester).

## 2.2.2. Preparation of Mangifera Indica Peels Catalyst

Unripe *Mangifera indica* fruit was carefully peeled and the peels were washed with ionized water, oven dried at 70 °C for 38 h until the constant weight was achieved. The dried peels were milled into powder and then sieved to obtain the fine powder. The fine powdered was placed in a crucible and heated to a temperature of 350 °C in a furnace for 4 h to achieved proper dryness. The dried *Mangifera indica* powdered was packed into a cleaned covered plastic bottle for further use as catalyst.

#### 2.2.3. Characterization of Mangifera Indica Powdered

The elemental contents of the *Mangifera indica* powdered was determined by using Atomic Absorption Spectrometry (AAS) for Magnesium, Flame photometer for Sodium and XRF spectrometry for other elements (aluminum, silicon, phosphorous, chlorine, potassium, calcium, zinc, iron, copper, chromium, manganese, nickel etc.).

## 2.2.4. Transesterification of Beniseedoil to Beniseed Methylester (BME)

One Liter glass reactor was charged with a known weight of the preheated Beniseed oil on a hot plate magnetic stirrer, a known amount of *Mangifera indica* powdered acting as catalyst was dissolved in a specific volume of methanol in a separate 500 ml reactor flask, stirrer rigorously, and then transferred into the preheated oil placed on the hot plate magnetic stirrer. The content of reactor was heated up to a known temperature until the reaction reaches completion at a required time. The content of the reactor was transferred into a 250 ml separating funnel, and allowed to separate within 12 h. Two layers was clearly noted, the lower layer which contained glycerol was tapped off from the bottom funnel, and the upper layer which contained biodiesel was washed with ionized water to removed untapped glycerol, residual catalyst and methanol present. The clearly washed light yellowish biodiesel was dried over calcium chloride powder to absorb the unwanted water, and then filtered to remove the deliquescence calcium chloride. The dried, light yellowish, low viscous BME was calculated using (Eq. (1)).

$$BME = \frac{weight of produced biodiesel}{weight of the preheated oil used}$$
(1)

## 2.2.4.1. Experimental Design for Transesterication and Statistical Analysis by RSM

To test the effects of reaction variables (reaction time, temperature of reaction, catalyst amount and methanol / oil molar ratio) RSM was employed. A central composite rotatable design

(CCRD) with four – factor – five – levels, which generate 30 experimental runs including factorial points, center points and axial points, was used for fitting a second order-four-variables response surface. Table 1 gives the factors and their levels while the 30 – CCRD experimental runs generated are presented in Table 2. The experiment was randomized to minimize the effects unexpected variability in the experimental responses. The simple linear regression model equation showing the functional relationship between the methylester yield and the independent variables is a linear function plus and error term is presented in (Eq. (2)).

$$Y = \alpha_o + \alpha_1 X_1 + \alpha_2 X_2 + \alpha_3 X_3 + \alpha_4 X_4 + \alpha_{12} X_1 X_2 + \alpha_{13} X_1 X_3 + \alpha_{14} X_1 X_4 + \alpha_{23} X_2 X_3 + \alpha_{24} X_2 X_4 + \alpha_{34} X_3 X_4 + \alpha_{11} X_1^2 + \alpha_{22} X_2^2 + \alpha_{33} X_3^2 + \alpha_{44} X_4^2 + \epsilon$$
(2)

 $\Upsilon$  is the value of response corresponding (biodiesel yield) to the value of  $X_1, X_2, X_3, X_4$  of the explanatory variable,  $\alpha_0$  is the intercept,  $\alpha_1, \alpha_2, \alpha_3$  and  $\alpha_4$  are the linear coefficients,  $\alpha_{12}, \alpha_{13}, \alpha_{14}, \alpha_{23}, \alpha_{24}$  and  $\alpha_{34}$  are interaction coefficients, while  $\alpha_{11}, \alpha_{22}, \alpha_{33}$  and  $\alpha_{44}$  are the quadratic coefficients.  $\varepsilon$  is the error term. For the regression analysis of the experimental data, design expert 9/DX9. Exe.Software was used. The results obtained were also analyzed using statistical model equations (Eq (3), Eq. (4) and Eq. (5)).

$$R^{2} = 1 - \sum_{i=1}^{n} \frac{(X_{i,cal} - X_{i,exp})^{2}}{(X_{avg,exp} - X_{i,exp})^{2}}$$
(3)

$$ADD = \left\{ \frac{1}{n} \left[ \sum_{i=1}^{n} \left( \frac{X_{i,exp} - X_{i,cal}}{X_{i,exp}} \right) \right] \right\}$$
(4)

$$RMSE = \sqrt{\frac{\sum(X_{i,cal} - X_{i,exp})}{n}}$$
(5)

Where *n* is the number of data set,  $X_{i,cal}$  is the calculated values,  $X_{i,exp}$  is the experimental values and  $X_{avg,exp}$  is the average experimental values.

Variable	Symbo					
	1	Coded factor levels				
		-2	-1	0	1	2
Reaction time (min) (RT)	$\mathbf{X}_1$	40	45	50	55	60
Heterogeneous catalyst amount (wt %) (HCA)	$X_2$	0.5	0.6	0.7	0.8	0.9
Temperature of reaction (°C) (TR)	$X_3$	40	45	50	55	60
Methanol / oil molar ratio (M/OMR)	$X_4$	3	4	5	6	7

Table-1.Factors and their levels for composite central rotatable design

#### 2.3. Modeling and Optimization Studies by ANN and RSM

In this study, a 2.5 CPC-X Software Neural Power new version was used. Multilayer Normal Feed Forward (MNFF) and Multilayer Full Feed Forward (MFFF) networks were used for predictive accuracy of oil and biodiesel yields. Search approach used was force approach. The two networks were trained by the Quick Prop (QP) learning algorithms and default stopping of 120,000 iterations. For the input layer, a total of four (4) layer numbers was used and the node number of input layer was four (4). For the output layer, Node Number was one (1), the transfer function was Tanh and the slope of transfer function and the hidden Layer was also one (1).

Meanwhile, the node number of output layer was twelve (12) with Tanh transfer function. The slope of transfer function was also one (Fig. 1). Similarly for RSM, the CCD data was divided into training and testing data sets. Twenty four (24) out of the thirty experimental obtained data were used as the training set while six (6) were used for the testing set. The test of significance,

Run	RT	HCA	TR	M/	EY(BME)	Predicted	Residual	Predicted	Residual
	(min)	(wt.%)	(°C)	ОМ	% (w/w)	value RSM	RSM	value	ANN
	. ,	. ,		R				ANN	
1	45	0.6	45	4	83.00	82.49	0.51	83.577	0.57704
2	55	0.6	45	4	85.70	86.53	-0.83	85.153	0.54714
3	45	0.8	45	4	88.56	88.40	0.16	87.724	0.83622
4	55	0.8	45	4	87.55	87.74	-0.19	88.325	0.77495
5	45	0.6	55	4	94.78	94.78	0.000	95.521	0.74070
6	55	0.6	55	4	67.60	68.15	-0.55	67.813	0.21272
7	45	0.8	55	4	92.27	92.20	0.069	92.155	0.11506
8	55	0.8	55	4	84.40	84.17	0.23	84.457	0.05665
9	45	0.6	45	6	94.78	94.78	0.000	95.134	0.35393
10	55	0.6	45	6	88.62	87.88	0.74	88.422	0.19777
11	45	0.8	45	6	71.10	71.00	0.10	70.996	0.10408
12	55	0.8	45	6	90.10	90.54	-0.44	90.304	0.20392
13	45	0.6	55	6	81.50	81.64	-0.14	81.409	0.09138
14	55	0.6	55	6	96.20	96.68	-0.48	95.406	0.79387
15	45	0.8	55	6	60.86	61.01	-0.15	60.898	0.03839
16	55	0.8	55	6	86.35	86.45	-0.10	86.44	0.09047
17	40	0.7	50	5	84.42	84.81	-0.39	84.504	0.08435
18	60	0.7	50	5	92.78	92.40	0.38	93.776	0.99637
19	50	0.5	50	5	94.78	94.78	0.000	94.463	0.31722
20	50	0.9	50	5	90.80	91.08	-0.28	91.127	0.32732
21	50	0.7	40	5	69.13	69.32	-0.19	69.133	0.00291
22	50	0.7	60	5	94.78	94.78	0.000	94.846	0.06647
23	50	0.7	50	3	93.30	92.08	1.22	92.619	0.68139
24	50	0.7	50	7	91.49	90.95	0.54	91.573	0.08283
25	50	0.7	50	5	70.50	70.03	0.47	88.903	18.4030
26	50	0.7	50	5	94.78	94.78	0.000	88.903	5.8772
27	50	0.7	50	5	94.78	94.78	0.000	88.903	5.8772
28	50	0.7	50	5	96.62	96.94	-0.32	88.903	7.7172
29	50	0.7	50	5	81.70	81.38	0.32	88.903	7.2028
30	50	0.7	50	5	95.59	96.27	-0.68	88.903	6.6872

Table-2.Central Composite Design (CCD), Experimental, Predicted and Residual Values for Five – Level-Four Factors Response Surface Analysis

ANOVA test and regression equation were used to achieve the modeling and optimization involved in BME production.

## 2.4. Analysis of Qualities of BME and There Free Fatty Acid Compositions

The qualities such as refractive index, moisture content, viscosity at 40 °C, acid value, saponification value, peroxide value, specific gravity, % FFA, cetane number etc. of BMEwas carried out by AOAC [22] methods, the mean molecular mass was obtained by the method of Akintayo and Bayer [23], the higher heating value was determined by Demirbas [24] method,

iodine value was obtained by Wijs method. The pour point and cloud point of the methylesterwere determined by the standard ASTM D2500 method. Free fatty acid of methylester was determined using gas chromatography (GC) (HP 6890 powered with HP Chem Station Rev. A 09.01 [1206] Software). 50 mg methylester was esterified for five minute at 95 °C with 3.4 ml of the 0.5 M KOH in dry methanol. The mixture was neutralized using by adding 0.7 M HCl and 3 ml of 14 % boron triflouride in methanol. The whole mixture was heated for 5 min at the temperature of 90 °C to achieve complete methylation process. The fatty acids were thrice extracted from the mixture with redistilled n-hexane. The content was concentrated to 1 µl for gas chromatography analysis with 1 µl was injected into the injection port of GC.

## 3. RESULTS AND DISCUSSION

# 3.1. Elemental Analysis of Mangifera Indica Powdered Catalyst

The results of elemental analysis of *Mangifera indica* powdered used as heterogeneous catalyst were presented in Table 3. Analyses showed that the *Mangifera indica* powdered consist of macro as well as micro elements. The macro elements were found to be K (59.85 %), Si (30.53 %), Cl (4.58 %), Al (3.05 %) and Ca (1.05 %). The major micro elements are P (0.196 %), S (0.593 %), Mn (0.043 %), Fe (0.037 %), Zn (0.008 %), Rb (0.042 %) and Sr (0.032 %). The catalytic activity of the *Mangifera indica* powdered can be traced to high proportion of macro elements and the problem of leaching of other metallic constituents such as Na and Ca into the biodiesel product was not observed as a result of low levels of these metals in the powder [18].

S/N	Elemental symbols	Concentration (g/l)	Percentage mass fraction %
1	Al	12,543.6	3.04732567
2	Si	125,643.4	30.52364219
3	Р	806.4	0.19590575
4	S	2440.1	0.59279468
5	Cl	18,872.2	4.58478742
6	К	246,342.6	59.84614693
7	Ca	4312.7	1.04772166
8	Mn	176.6	0.04290297
9	Fe	152.2	0.03697526
10	Zn	32.4	0.00787121
11	Rb	172.3	0.04185834
12	Sr	132.0	0.03206791
SUM		411,626.5	100

Table-3. Elemental analysis of Mangifera indica peels powdered

#### 3.3. Modeling and Optimization of Transesterification Reaction by RSM

The results of the transesterification reaction using CCRD experimental design was presented in Table 2 above. Fig 1 showed the plot of predicted against the experimental biodiesel yield. Based on the constraints and the single objective function in (Eq. (2)), the model equation

that correlate the controllable variables and the single objective function (response BME) is given in Eq. ((6)).

$$BME \ \%/= 94.78 + 2.22x_1 + 3.87x_2 + 2.56x_3 + 6.32x_4 - 1.25x_1x_2 + 1.05x_1x_3 - 2.09x_1x_4 - 1.37x_2x_3 - 4.25x_2x_4 - 0.70x_3x_4 - 2.17x_1^2 - 4.43x_2^2 - 0.80x_3^2 - 2.79x_4^2$$

$$(6)$$

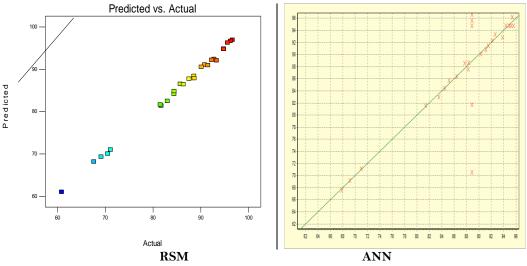


Fig-1.Graph of predicted against the Actual yield for RSM and ANN

ANOVA test, test of significant, coefficient of determination ( $\mathbb{R}^2$ ) and  $R_{Adj}^2$ , were used to evaluate the statistical significant of the model equation. The results of test of significant test were presented in Table 4. Observation on the results showed that all the variables are remarkable significant at 95% CI level (p<0.05). However, the variable X<sub>4</sub> with *F*-test value = 2621.12, p<0.0001, is the most significant variable. The results of ANOVA and analysis of variance were presented in Table 5. Observation showed that all the factors has negative coefficient of estimation effects on the response except X<sub>1</sub>, X<sub>2</sub>, X<sub>3</sub>, X<sub>4</sub> and X<sub>1</sub>X<sub>3</sub>, respectively with positive impact.

All the degree of fredom (df) associated with sum of the squares in Table 5 are unity (1), the standard error are all lesser than one, and the variance inflation factor ranged from 1.00 to 1.05 implied a high significance for the regression model [25]. The coefficient of determination ( $\mathbb{R}^2$ ) and  $R_{Adj}^2$  were evaluated as 98.85% and 99.62%, respectively.

Source	Sum of Squares	df	Mean Square	F-Value	p-value
$X_1$	118.77	1	118.77	324.88	< 0.0001
$X_2$	359.99	1	359.99	984.69	< 0.0001
$X_3$	156.72	1	156.72	428.69	< 0.0001
$X_4$	958.24	1	958.24	2621.12	< 0.0001
$X_1X_2$	24.88	1	24.88	68.04	< 0.0001

Table-4. Test of Significance for Every Regression Coefficient of CCRD

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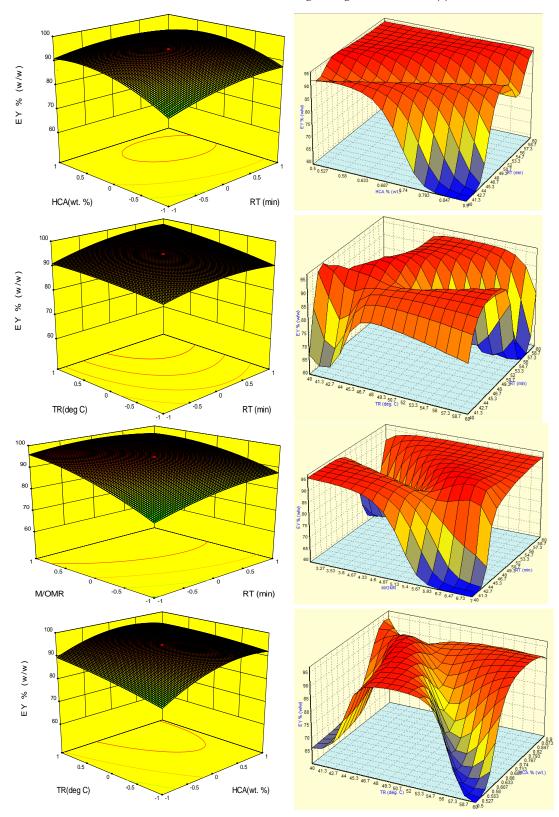
$X_1X_3$	17.62	1	17.62	48.19	< 0.0001
$X_1X_4$	69.93	1	69.93	191.29	< 0.0001
$X_2X_3$	29.89	1	29.89	81.77	< 0.0001
$X_2X_4$	289.60	1	289.60	792.14	< 0.0001
$X_3X_4$	7.85	1	7.85	21.48	0.0003
$X_{1^2}$	129.44	1	129.44	354.07	< 0.0001
$X_{2^2}$	537.95	1	537.95	1471.49	< 0.0001
$X_{3^{2}}$	17.71	1	17.71	48.46	< 0.0001
$X_4^2$	212.92	1	212.92	582.40	< 0.0001

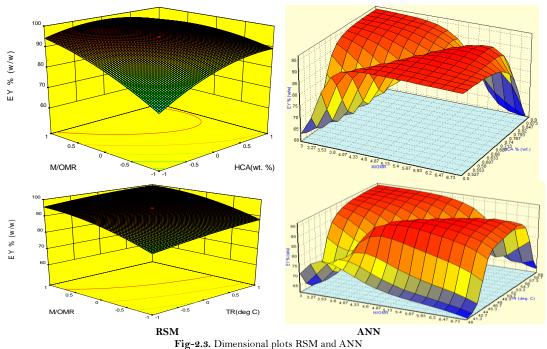
Factors	Coefficient Estimate	df	Standard Error	95%CI Low	95%CI High	VIF
Intercept	94.78	1	0.25	94.25	95.31	-
X <sub>1</sub>	2.22	1	0.12	1.96	2.49	1.00
$X_2$	3.87	1	0.12	3.61	4.14	1.00
$X_3$	2.56	1	0.12	2.29	2.82	1.00
$X_4$	6.32	1	0.12	6.06	6.58	1.00
$X_1X_2$	-1.25	1	0.15	-1.57	-0.92	1.00
$X_1X_3$	1.05	1	0.15	0.73	1.37	1.00
$X_1X_4$	-2.09	1	0.15	-2.41	-1.77	1.00
$X_2X_3$	-1.37	1	0.15	-1.69	-1.04	1.00
$X_2X_4$	-4.25	1	0.15	-4.58	-3.93	1.00
$X_3X_4$	-0.70	1	0.15	-1.02	-0.38	1.00
$X_{1^{2}}$	-2.17	1	0.12	-2.42	-1.93	1.05
$X_{2^2}$	-4.43	1	0.12	-4.67	-4.18	1.05
$X_{3}^{2}$	-0.80	1	0.12	-1.05	-0.56	1.05
$X_4^2$	-2.79	1	0.12	-3.03	-2.54	1.05
Analysis of Variand	e of Regression E	quation		•	•	•
Source	Sum of squares	df	Mean Square	<b>F-value</b>	p-value	
Model	2748.86	14	196.35	537.08	< 0.0001	
Residual	5.48	15	0.37			
Lack of Fit	5.48	10	0.55			
Pure Error	0.000	5	0.000			
Cor Total	2754.34	29				
Variance						1.858
Standard deviation						1.363
$R^2 = 98.85\%,$	$R^{2}_{adj}-Sq = 99.62\%$	• )	•	•		

Table-5.ANOVA for Response Surface Quadratic Model for Intercept.

The variance activity that varies from a norm was determined to be 1.858, while the square root of the variance called the standard deviation was obtained to be 1.363.

Fig. 2, showed the 3-D graphical representation of response. The curvature nature of the graphs proved that there are perfect interactions among the controllable variables  $(X_1, X_2, X_3 \text{ and } X_4)$ . To establish the optima condition, Forty two (42) variables combination were generated, the optimal





0

Variables condition for this process was established at thirty one (31) variable combinations. This was obtained by minimizing the controllable variables while maximizing the response variable (BY). The condition was established at  $X_1 = 62.0 \text{ min}$ ,  $X_2 = 0.9 \text{ wt. }$ %,  $X_3 = 60^{\circ}\text{C}$  and  $X_4 = 6.5$ . The predicted BME under this condition was 94.20 % (w/w). Using these optimal condition values for three independent replicates, a mean of 92.80 % (w/w) BME was achieved, which was within the range predicted by the model.

## 3.3. Modeling and Optimization of Transesterification Reaction by ANN

The experimental results, the predicted and the residual values were presented in Table 2 above. However, Fig. 1 above, showed the graph of predicted against the experimental yield. Based on the force approach, the connection value of N3L1-N3L2 [3.9554], with the lower [2.9554] and upper value [4.9554] proved suitable for determination of the root mean square error (RMSE) as well as average coefficient of determination ( $\mathbb{R}^2$ ) and  $R^2_{Adj}$ . The RMSE,  $\mathbb{R}^2$  and  $R^2_{Adj}$ , were evaluated to be 0.05100, 99.999% and 99.997%, respectively. The variance activity that varies from a norm was determined as 53.8600, while the square root of the variance called the standard deviation was obtained to be 7.33894.

Depict also in Fig. 2 above is the 3 - dimensional graphical representation by ANN software. The nature of the graphs indicated that there are outstanding mutual interactions among the variables. To establish the optima condition, forty seven (47) variables combination were generated, the optimal variables condition for this process was established at thirty three (33) variables combination. This was obtained by minimizing the controllable variables while maximizing the response variable (BME). The condition was established at  $X_1$ = 60.0 min,  $X_2$  = 1.0 wt. %,  $X_3$ = 57°C and  $X_4$  = 6.0. The predicted BME under this condition was 94.40 % (w/w). Using these optimal condition values for three independent replicates, a mean of 93.80 % (w/w) BME was achieved, which was well within the range predicted by the model.

#### 3.4. ANN and RSM Performance Comparison of Beniseed Methylester (BME)

The accuracy of models (ANN and RSM) was checked by comparing the graph of the predicted and the actual BME. Meanwhile, AAD, RMSE,  $R^2$  and  $R^2_{Adj.}$  variance and standard deviation were also evaluated to show the superiority of one model over the other. The graph of the predicted against the actual value showed by ANN indicted that nearly all of the predicted value are well within the actual value. The evaluated results of AAD, RMSE,  $R^2$  and  $R^2_{Adj.}$  variance and standard deviation revealed that the ANN and RSM are good optimization tools with accurate predictions (Table 5). Nevertheless, the results displayed the supremacy of ANN over RSM for the reason that predicted values (93.60 % w/w for ANN; 92.80 %w/w for RSM),  $R^2$  (99.999% for ANN; 99.44% for RSM) and  $R^2_{Adj.}$ (99.997% for ANN; 98.72% for RSM). It can be therefore concluded that artificial neural network proved to be better in terms of modeling and optimization of transesterification of Beniseed oil to BME.

Data	Values				
	RSM	ANN			
AAD	0.2100	0.1820			
RMSE	0.04567	0.005196			
$\mathbb{R}^2$	0.9944	0.99999			
Adjusted R <sup>2</sup>	0.9872	0.99997			
Variance	1.3630	53.8600			
Standard deviation	1.2767	7.33894			
Predicted BME %(w/w)	92.80	93.60			

Table-6.Comparism of ANN with RSM

# 4. CONCLUSION

This study demonstrated that Beniseed oil can be easily transesterified to Beniseed methyl ester using a heterogeneous base catalyst. The transesterification process was effectively optimized by statistical software (ANN and RSM). The study displayed that ANN perform better than RSM due to comparison of predicted value, RMSE, AAD, R<sup>2</sup>, R<sup>2</sup><sub>Adj</sub>. In the meantime, the fatty acid profile and the physicochemical properties of the BME proved that, BME can serve as alternative fuel for conventional diesel.

## 5. ACKNOWLEDGEMENTS

The effort of Engr. Adesina Olusola, the staff of Landmark University, Omu-Aran, Kwara State, Nigeria, is deeply appreciated.

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